

35, and 50 kV. The variation observed was less than the standard deviation of the measurements. For the 444 reflection with Cu radiation at 20, 35 and 50 kV, there was no difference in the values of a , the difference in measured θ_{444} being only 1.0 sec.

In the same way, and using the same germanium crystal, determination was made of the ratio of the Co $K\alpha_1$ wavelength and of the Fe $K\alpha_1$ wavelength to Cu $K\alpha_1$. λ Co $K\alpha_1$ was computed from the 333 reflection for which $\theta = 55.2^\circ$. λ Fe $K\alpha_1$ was measured from the 333 reflection at $\theta = 62.7^\circ$. A summary of the calculation and corrections is given in Table 1. The calculated wavelengths and wavelength ratios are shown in Table 2 where they are compared with the values of Siegbahn.

Table 2. Calculated wavelengths and ratios

	Cooper	Siegbahn
Cu $K\alpha_1$	1.537395 kX (assumed)	1.537395 kX
Mo $K\alpha_1$	0.707840 ± 0.000005	0.707831
Co $K\alpha_1$	1.78528 ± 0.00002	1.78529
Fe $K\alpha_1$	1.93204 ± 0.00001	1.932076
$\lambda(\text{Mo})/\lambda(\text{Cu})$	0.460415	0.460409
$\lambda(\text{Co})/\lambda(\text{Cu})$	1.16124	1.16124
$\lambda(\text{Fe})/\lambda(\text{Cu})$	1.25670	1.25672

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New intermediate phases in transition metal systems. II. By B. C. GIessen and N. J. GRANT, *Department of Metallurgy, Massachusetts Institute of Technology, Cambridge, Massachusetts, U.S.A.*

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This report is a continuation of a former abstracted communication (Giessen & Grant, 1964) on intermediate phases encountered in the course of an examination of binary transition metal systems containing a refractory metal component. Detailed descriptions of this work have been published, or submitted for publication (Giessen, Ibach & Grant, 1964; Ritter, Giessen & Grant, 1964; Giessen, Kane & Grant, 1965). Since, additional binary and ternary systems

The Co/Cu ratio is identical with Siegbahn's, while the Fe/Cu ratio differs by 16 ppm, and the Mo/Cu ratio differs by 13 ppm.

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have been investigated and 18 new ordered phases have been found. Detailed descriptions of this work will again be published elsewhere.

The structures were all derived from powder patterns taken with an XRD 5 diffractometer or with a 114.6 mm camera and Cu $K\alpha$ radiation; this was made possible by the fact that the complex close-packed structures are related, and that their powder patterns show typical coincid-

Table 1. New binary and ternary transition metal phases

Compositional designation	Crystal system	Structure type	Layer structure symbol	a_0	b_0	c_0
MoIr	Orthorhombic	B19-AuCd	AB(2)	2.752 Å	4.804 Å	4.429 Å
WIr	Orthorhombic	B19-AuCd	AB(2)	2.760	4.811	4.452
α -VIr	Orthorhombic	α -VIr	—	5.791	6.756	2.796
$(\text{V}_{0.96}\text{Ir}_{0.04})\text{Ir}:(\beta\text{-VIr})$	Tetragonal	L1 ₀ -AuCu	AB(3)	3.887	—	3.651
TaNi ₂	Tetragonal	MoSi ₂	—	3.154	—	7.905
MoIr ₃	Hexagonal	DO ₁₉ -MgCd ₃	AB ₃ (2) sh 0	5.487	—	4.385
WRh ₃	Hexagonal	DO ₁₉ -MgCd ₃	AB ₃ (2) sh 0	5.453	—	4.350
β -TaNi ₃	Monoclinic	β -NbPt ₃ *	AB ₃ (12) sh $\frac{1}{2}$	5.11	4.54 $\alpha = 90^\circ 38'$	25.5 ₀
Ta(Pd _{0.92} Rh _{0.08}) ₃	Orthorhombic	β -NbPd ₃ †	AB ₃ (6) sh $\frac{1}{2}$ ‡	5.492	4.829	13.54
Ta(Pd _{0.72} Rh _{0.28}) ₃	Rhombohedral	β -Ta(Pd, Rh) ₃ (ordered Sm)	AB ₃ (9) sh 0	5.517	—	20.26
Ta(Pd _{0.88} Ru _{0.12}) ₃	Rhombohedral	β -Ta(Pd, Rh) ₃	AB ₃ (9) sh 0	5.531	—	20.25 ₂
Ta(Pd _{0.67} Rh _{0.33}) ₃	Hexagonal	γ -Ta(Pd, Rh) ₃	AB ₃ (10) sh 0	5.52 ₀	—	22.4 ₃
Ta(Pd _{0.5} Rh _{0.5}) ₃	Hexagonal	PuAl ₃ (VCO ₃)	AB ₃ (6) sh 0	5.530	—	13.49 ₃
Ta(Pd _{0.75} Ru _{0.25}) ₃	Hexagonal	PuAl ₃ (VCO ₃)	AB ₃ (6) sh 0	5.509	—	13.44
Ta(Pd _{0.725} Rh _{0.225}) ₃ (H.T.)	Hexagonal	DO ₂₄ -TiNi ₃	AB ₃ (4) sh 0	5.517	—	8.978
Ta(Pd _{0.78} Ru _{0.22}) ₃ (H.T.)	Hexagonal	DO ₂₄ -TiNi ₃	AB ₃ (4) sh 0	5.525	—	9.027
Ta(Pd _{0.5} Ru _{0.5}) ₃	Cubic	L1 ₂ -AuCu ₃	AB ₃ (3) sh 0	3.893	—	—
(Ta _{0.5} Al _{0.5})Ni ₃	Hexagonal	DO ₂₄ -TiNi ₃	AB ₃ (4) sh 0	5.112	—	8.340

* Giessen, Kane & Grant (1965).

† Giessen & Grant (1964).

‡ Shift density not certain.

Table 2. *New structure types* α -VIrOrthorhombic, space group $Cmmm$ (D_{2h}^{19}); $N=8$ atoms per cell.V in $4(g) \pm (x, 0, 0)$; $x \sim 0.28$;
Ir in $4(j) \pm (0, y, \frac{1}{2})$; $y \sim 0.22$. β -Ta(Pd, Rh)₃Rhombohedral, space group $R\bar{3}m$ (D_{3d}^5); $N=36$ atoms per hexagonal cell; stacking sequence: $ababcacac$, $ababcacac$.Ta in $3(a) (0, 0, 0)$; $6(c) \pm (0, 0, z)$; $z_c \sim \frac{2}{3}$;Pd_{0.72}Rh_{0.28} in $9(e) (\frac{1}{2}, 0, 0)$; $0, \frac{1}{2}, 0$; $\frac{1}{2}, \frac{1}{2}, 0$; $18(h) \pm (x, \bar{x}, z)$; $x, 2x, z$; $2\bar{x}, \bar{x}, z$;
 $x \sim \frac{1}{2}$; $z_h \sim \frac{3}{8}$. γ -Ta(Pd, Rh)₃Hexagonal, space group $P6_3/mmc$ (D_{6h}^4); $N=40$ atoms per unit cell; stacking sequence: $abcacacbc$, $abcacacbc$.Ta in $2(b) \pm (0, 0, \frac{1}{2})$; $2 \times 4(f) \pm (\frac{1}{3}, \frac{2}{3}, z)$; $z_{f1} \sim \frac{3}{8}$; $z_{f2} \sim -\frac{1}{8}$;Pd_{0.67}Rh_{0.33} in $6(h) \pm (x, 2x, \frac{1}{4})$; $2\bar{x}, \bar{x}, \frac{1}{4}$; $x, \bar{x}, \frac{1}{4}$; $2 \times 12(k) \pm (x, 2x, z)$; $2\bar{x}, \bar{x}, z$; x, \bar{x}, z ; $\bar{x}, 2\bar{x}, \frac{1}{2} + z$; $2x, x, \frac{1}{2} + z$; $\bar{x}, x, \frac{1}{2} + z$; $x_{k1} \sim -\frac{1}{8}$; $z_{k1} \sim \frac{3}{8}$; $x_{k2} \sim \frac{1}{8}$; $z_{k2} \sim -\frac{3}{8}$.

ing lines. Error limits are $< 5 \cdot 10^{-4}$, except for β -TaNi₃ and Ta(Pd_{0.67}Rh_{0.33})₃, where they are $2 \cdot 10^{-3}$, and $1 \cdot 10^{-3}$, respectively. Table 1 lists the new intermediate phases, including again a layer structure symbol as introduced previously (Giessen & Grant, 1964). Table 2 presents three new structure types.

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MX₄ compounds of alkaline earth metals with IIIB group elements. By GIACOMO BRUZZONE, *Institute of Physical Chemistry, Genoa University, Genoa, Italy**

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In the course of a study on the equilibrium diagrams of alkaline earth metals with the elements of the IIIB group of the periodic system, intermediate phases of the type MX₄ have been found in the systems Ca-Ga, Sr-Ga, Ba-Ga, Ba-In, Ba-Tl.

The gallium alloys were prepared by fusion under argon atmosphere in Phytogoras and those of indium and thallium in sealed iron crucibles. The chemical analysis showed that the compositions of the specimens were in agreement ($\pm 1\%$) with the stoichiometric values. The (Ca, Sr, Ba)Ga₄ compounds are grey and brittle; BaIn₄ and BaTl₄ are white and rather soft. These MX₄ phases, except BaTl₄ (whose different structure type was not determined), show the tetragonal D_{13} structure type ($I4/mmm - D_{4h}^{17}$) known so far for aluminum compounds (Andress & Alberti, 1935; Nowotny & Wesenberg, 1939; Nowotny, Wormnes & Mohrnheim, 1940, 1942). The atomic positions are: $[0, 0, 0; \frac{1}{2}, \frac{1}{2}, \frac{1}{2}] + : 2M(0, 0, 0); 4X' \pm (0, 0, z); 4X''(0, \frac{1}{2}, \frac{1}{2})(\frac{1}{2}, 0, \frac{1}{2})$; (M = alkaline earth atom). The unit cell is shown in Fig. 1.

The powder patterns obtained with the Straumanis method using copper $K\alpha$ radiation gave the values of the lattice constants recorded in Table 1. The best agreement of the observed intensities with those calculated for z between

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In addition to the phases listed in Table 1, ordered WIr₃ of the MgCd₃ type probably exists, but this could not be proved conclusively owing to the similar scattering powders of W and Ir. The structure of α -VIr is related to $B2$ -CsCl by a distortion of the atom positions and approximate doubling of two $B2$ -cell edges. Most of the ternary AB₃ phases of the types Ta(Pd_{*x*}Rh_{*1-x*})₃ and Ta(Pd_{*x*}Ru_{*1-x*})₃ are related in such a fashion that a similar valence electron concentration in both series causes identical structures to appear; in this context the occurrence of Ta(Pd_{0.5}Ru_{0.5})₃ with the $L1_2$ -AuCu₃ structure of TaRh₃ is remarkable. The significance of the valence electron concentration in relation to the structure type is also borne out by the occurrence of (Ta_{0.5}Al_{0.5})Ni₃ with the TiNi₃ structure, where equal parts of Ta and Al substitute for Ti.

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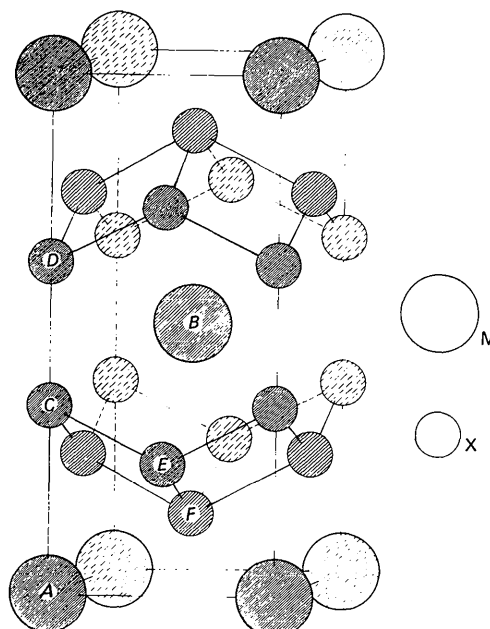


Fig. 1. Unit cell of MX₄ (M = Ca, Sr, or Ba; X = Ga, In or Tl).